Practical Issues in Using Force Fields

Computational Bottlenecks

- The evaluation of U_{local} , with n atoms: each atom type is involved in $x_i < k$ bonds, so it takes O(kn) = O(n) linear
- ullet Conversely, the evaluation of $U_{nonlocal}$ takes $O(n^2)$ quadratic
- To reduce computational cost, three techniques have been developed:
 - Spherical cutoff
 - Particle-mesh Ewald method (PME)
 - Multipole schemes

Hofstadter's law:

It always takes longer than you expect, even when you take into account Hofstadter's law (D.R. Hofstadter, in *Gödel*, *Escher*, *Bach*, 1979)

Spherical Cutoff

- Cutoff radius b: For r>b, always U=0
- Each atom, within a distance **b**, has at least other $x_i < k$ atoms, so the evaluation of $U_{nonlocal}$ takes O(kn) = O(n) linear
- Cutoff can be used either for energy or force functions
 - Truncation: U unaltered for r<b, 0 for r>b
 - Switching: for r in [a,b], U goes to 0
 - Shift: U gradually changes for each r<b/i>
- Usually, in Å, [a,b]≈[8,12] or [a,b]≈[11,15]
- Choice criteria:
 - Short-range energies/forces possibly kept unaltered
 - Energies altered gradually (avoid spurious minima)
 - Cutoff must avoid spurious large forces around b
 - Energy should be conserved...

General Cutoff Formulation

• The general modification can be expressed as follows:

$$U_{non-bonded} = \sum_{ij} w_{ij} \, {\color{red} S(r_{ij})} \, [(-A_{ij}/r_{ij}{}^6 + B_{ij}/r_{ij}{}^{12}) + \, q_i q_j/r_{ij}]$$
 (between atoms i and j)

- w_{ij} are weights ($0 \le w_{ij} \le 1$) possibly used to exclude bonded or bond/angle terms, etc.
- Depending on the particular S(r), different approximations apply.

Truncation function S(r):
$$S(r) = \begin{cases} 1 & r < b \\ 0 & r \ge b \end{cases}$$

Potential Switch

$$U_{non-bonded} = \sum_{ij} w_{ij} S(r_{ij}) \left[(-A_{ij}/r_{ij}^{6} + B_{ij}/r_{ij}^{12}) + q_{i}q_{j}/r_{ij} \right]$$

- In S(r), the polynomial degree must be sufficiently high to assure that energy and its gradient are continuous functions.
- **Potential Switch** function S(r) (example):

$$S(r) = \begin{cases} 1 & r < a \\ 1 + y(r)^{2}(2r - 3) & a \le r < b \\ 0 & r > b \end{cases}$$
with
$$y(r) = (r^{2} - a^{2})/(b^{2} - a^{2})$$

Region [a,b] is known as the switching buffer region

Shift Functions

$$U_{non-bonded} = \sum_{ij} w_{ij} S(r_{ij}) \left[(-A_{ij}/r_{ij}^{6} + B_{ij}/r_{ij}^{12}) + q_{i}q_{j}/r_{ij} \right]$$

- Shift functions avoid abrupt changes in forces, but lead to underestimates of short-range forces.
- **Potential Shift** functions S(r) (example):

$$S_{1}(r) = \left[1 - \left(\frac{r}{b}\right)^{2}\right]^{2} \qquad for \quad r \leq b$$

$$or$$

$$S_{2}(r) = \left[1 - \frac{r}{b}\right]^{2} \qquad for \quad r \leq b$$

• Previous functions are usually applied to electrostatic potential.

"Additive" Shift

$$U_{Van \ der \ Waals} = \sum_{ij} w_{ij} \left[(-A_{ij}/r_{ij}^{6} + B_{ij}/r_{ij}^{12}) + S(r_{ij}) \right]$$

- The Van der Waals potential is usually shifted to zero by an additive term (e.g. in CHARMM)
- Additive Shift function S(r) (example):

$$\overline{U}_{VdW}(r) = \begin{cases} \sum_{i,j} w_{ij} \left[\left(-\frac{A_{ij}}{r_{ij}^{6}} + \frac{B_{ij}}{r_{ij}^{12}} \right) + S_{ij}(r) \right] & for \quad r \leq b \\ 0 & for \quad r > b \end{cases}$$

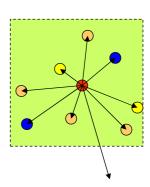
with

$$S_{ij}(r) = C_{ij}r_{ij}^6 + D_{ij}$$

Constants C and D are chosen so that both potential and force are 0 at b.

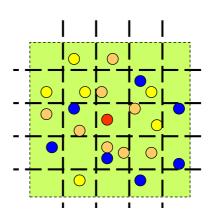
Interaction Computation

- The computation of interactions could consider all possible pairs but, if cutoff is adopted, data must be properly organized to exploit the trick.
 Possible methods:
- Cell subdivision
- Neighbor list



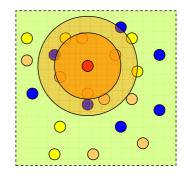
Cell Subdivision

- In "cell subdivision" the whole system is subdivided in a lattice of boxes whose edge is greater than the cutoff radius.
- Interactions can be checked only among adjacent cells.
- Data of each atom is kept in the list corresponding to the current hosting box
- As an atom enters a different cell, its data must be moved to another list



Neighbor List

- For each atom, we can keep a list of other atoms within a radius b+Δr
- The method success depends on the fact that such list remains valid over several timesteps (about 10-20) because of Δr...)
- The decision to refresh the lists depends on the monitored maximum velocity at each step



QSAR: What's This?

- QSAR stands for Quantitative Structure-Activity Relationships
- QSAR is a catch-all term in Chemistry that refers to (computational) methods aimed at obtaining quantitative results on chemical activity from molecular structure information
- 3D-QSAR is the application of force-fields calculations to achieve typical QSAR goals